


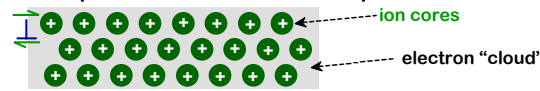
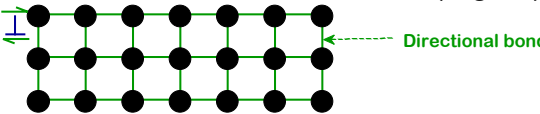
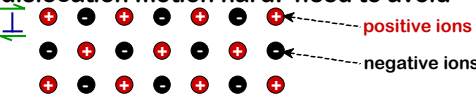
Chapter 7: Dislocations and Strengthening

ISSUES TO ADDRESS...

1. Why are dislocations observed primarily in metals and their alloys?
2. How are strength and dislocation motion related?
3. How do we increase strength?
4. How can heating change strength and other properties?

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Dislocation motion and material classes

- **Metals:** Dislocation motion easier:
 - nondirectional bonding
 - close-packed directions for slip
- **Covalent Ceramics: (Si, diamond)**
 - dislocation motion hard: directional (angular) bonding
- **Ionic Ceramics: (NaCl)**
 - dislocation motion hard: need to avoid ++ and -- neighbors

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Dislocation motion...

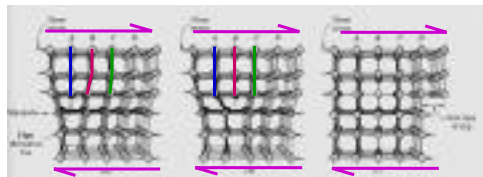
- Produces plastic deformation
- Depends on incrementally breaking bonds

stretched zinc crystal



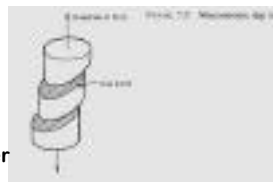
Callister Fig. 7.9


Callister Fig. 7.1



- If dislocations don't move, plastic deformation doesn't happen!

Callister Fig. 7.8



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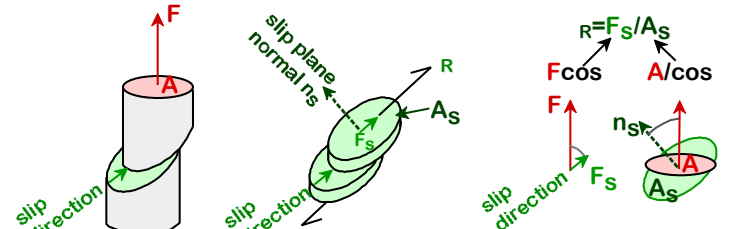
Stress and Dislocation Motion

- Crystals slip due to a resolved shear stress
- Applied tension can produce a resolved shear stress


Applied tensile stress: $\sigma = F/A$

Resolved shear stress: $\tau = F_s/A_s$

Relation between σ and τ



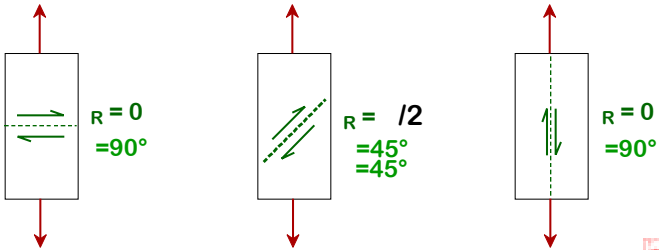
$$\tau = \sigma \cos \theta \cos \phi$$

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Critical Resolved Shear Stress: τ_{CRSS}

- Condition for disl. motion: $\tau > \tau_{CRSS}$
- Crystal orientation can make it easy or hard to move disl.
 - typically $10^{-4}G$ to $10^{-2}G$

$$R = \cos \phi \cos \lambda$$



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Dislocation motion in polycrystals

- Slip planes and directions (ϕ, λ) change from one crystal to another.
- R will vary from one crystal to another.
- Crystal with largest R yields first.
- Other (less favorably oriented) crystals yield later.



Callister Fig. 7.10

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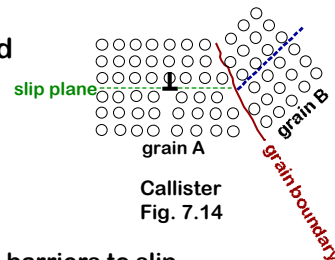


Strategy for Strengthening: Make dislocations hard to move

4 ways to do this:

1. Reduce the grain size, d

- Grain boundaries are barriers to slip.
- Barrier "Strength" increases with misorientation
- Smaller grain size: more barriers to slip



Callister Fig. 7.14

Hall-Petch Equation:

$$\sigma_{yield} = \sigma_0 + k_y d^{-1/2}$$

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Example of grain size strengthening

- 70wt%Cu-30wt% Zn brass alloy



Fig. 4.11(b) Callister

40 μ m

- yield = $\sigma_0 + k_y(\text{grain size})^{-1/2}$

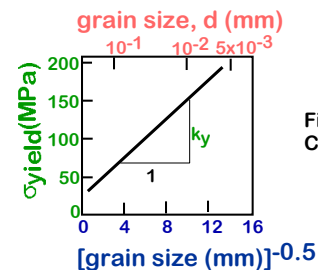


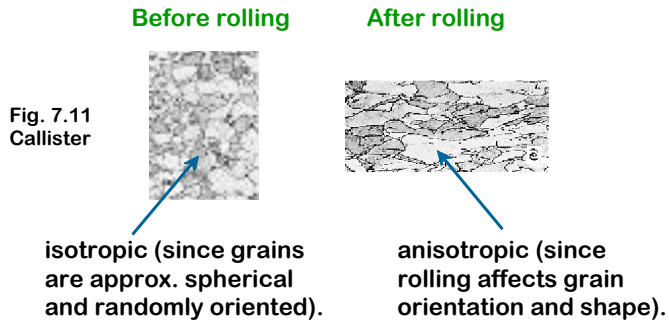
Fig. 7.15 Callister

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Anisotropy in yield

- Can be induced by rolling a polycrystalline metal

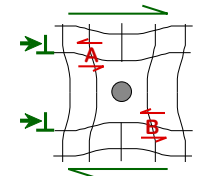


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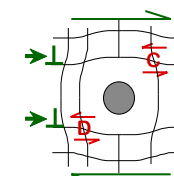


2. Solid Solution Strengthening

- Impurity atoms generate stress by distorting the lattice.
- This stress can produce a barrier to dislocation motion.
- Example: smaller substitutional impurity
- Example: larger substitutional impurity



Impurity generates local shear at **A** and **B** that opposes disl. motion to the right.



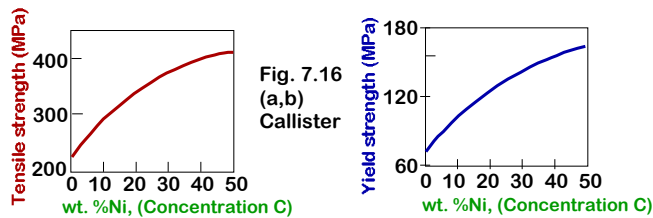
Impurity generates local shear at **C** and **D** that opposes disl. motion to the right.

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Ex: Solution Strengthening in Copper

- Tensile strength and yield strength increase with wt%Ni



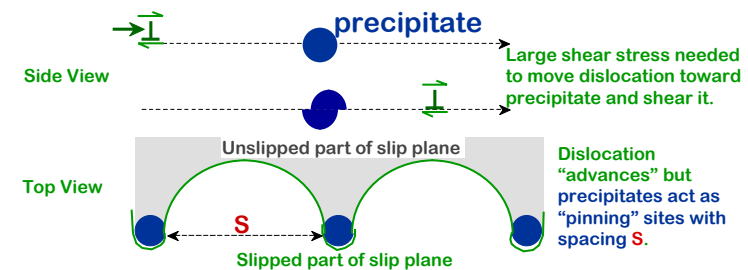
- Empirical relation: $\sigma_y \propto C^{1/2}$
- Alloying increases σ_y and TS

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3. Precipitation Strengthening

- Hard precipitates are difficult to shear
Ex.: ceramics in metals
(SiC precipitates in iron or in aluminum)



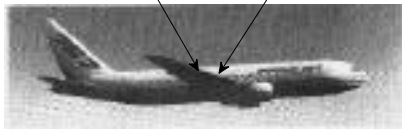
- Contribution to yield scales as: $1/S$

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Example: Precipitation-strengthened alloy

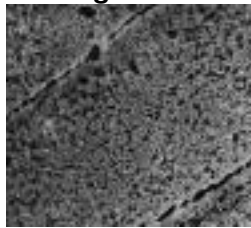
internal wing structures upper wing skins



Boeing 767

Callister, Fig. 11.0

- Aluminum is strengthened with precipitates formed by alloying.



1.5 μm

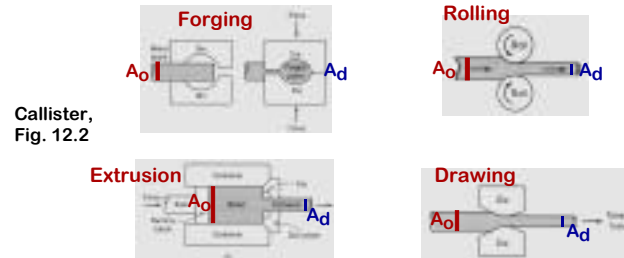
Callister, Fig. 11.0

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4. Cold Work (%CW)

- Room Temperature Deformation
- Common forming operations change the cross sectional area.



Callister, Fig. 12.2

$$\%CW = \frac{A_0 - A_d}{A_0} \times 100$$

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Dislocations during cold work

- Titanium alloy after cold working



Callister, Fig. 4.6

120 μm

- Dislocations entangle with one another during cold work.

- Dislocation motion becomes more difficult.

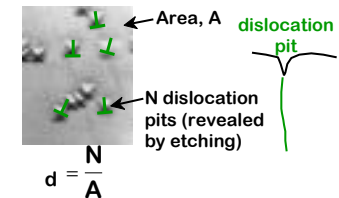
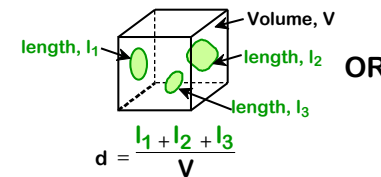
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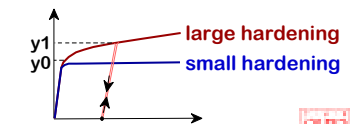
Result of Cold Work

Material condition	dislocation density (d)	d [mm/mm ³] (typical values)
Carefully prepared		10 ³
Heavily deformed		10 ¹⁰

- Ways of measuring d:



- Yield stress as d (i.e., the material hardens)

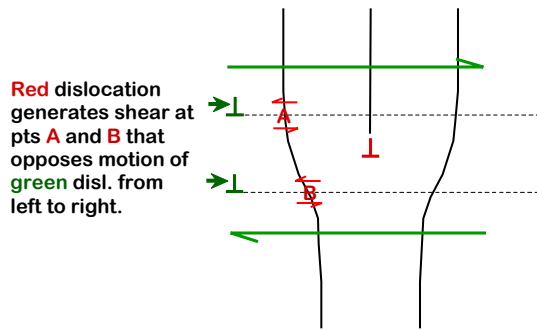


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Dislocation-dislocation trapping

- Dislocations generate stress
- This stress traps other dislocations

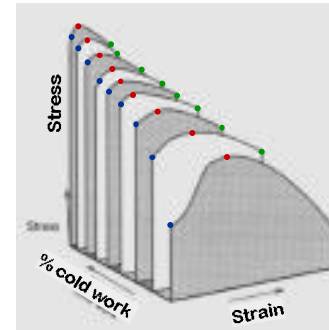


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Impact of Cold Work

- Yield strength (σ_y) increases
- Tensile strength (TS) increases
- Ductility (%EL or %AR) decreases



Callister, Fig. 7.20

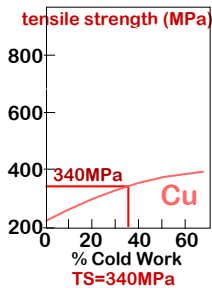
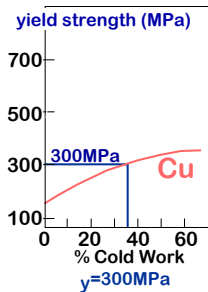
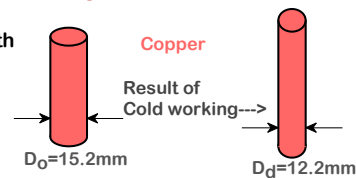
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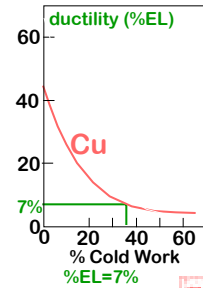
Cold Work Analysis

Question: what is the tensile strength and ductility after cold working?

$$\%CW = \frac{r_o^2 - r_d^2}{r_o^2} \times 100 = 35.6\%$$



Callister, Figs. 7.20(a,b,c)

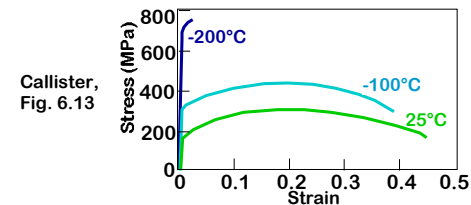


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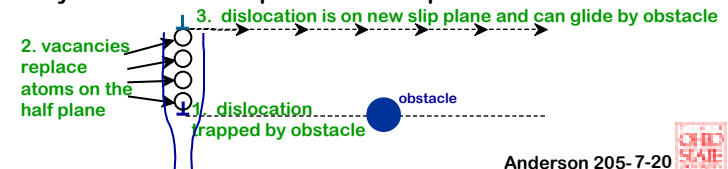
Behavior at different temperatures

- Results for polycrystalline iron:



Callister, Fig. 6.13

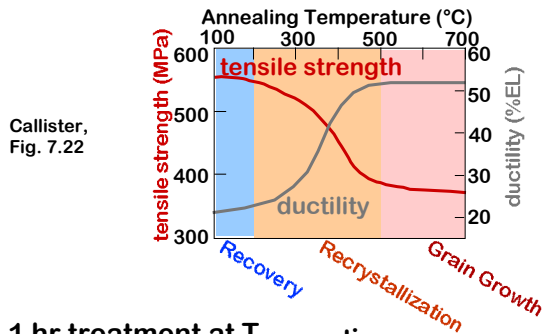
- σ_y and TS decrease with increasing test temperature.
- %EL increases with increasing test temperature.
- Why? Vacancies help dislocations past obstacles.



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Effect of heating after cold work



- 1 hr treatment at T_{anneal} :
 - decreases TS
 - increases %EL
- Effects of cold work are reversed!
- 3 Annealing stages to discuss...

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Recovery:

Annihilation reduces dislocation density

- Scenario #1
- Scenario #2

3. "Climbed" disl. can now move on new slip plane
2. grey atoms leave by vacancy diffusion allowing disl. to "climb"
1. dislocation blocked; can't move to the right
4. opposite dislocations meet and annihilate

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Recrystallization

- formation of new crystals that
 - have small d
 - are small
 - consume cold-worked crystals
- brass

33% cold worked



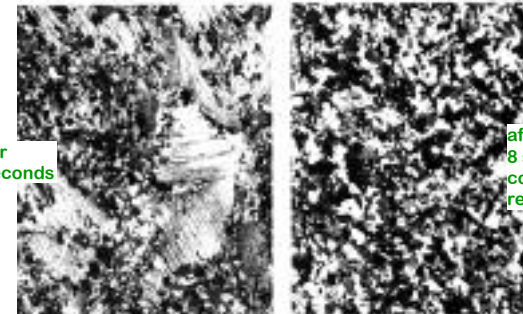
Callister, Fig. 7.21(a,b)

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Further Recrystallization

after 4 seconds



after 8 seconds - completely recrystallized!

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Grain Growth

- Larger grains consume smaller grains
- Why? Grain boundary area is reduced.



after 8 seconds
(580°C)



after 15 minutes
(580°C)

- empirical relation

$$d^n - d_0^n = Kt$$

grain diameter at time=t → d^n

grain diameter at t=0 → d_0^n

exponent: material-dependent (typ. ~2)

elapsed time → t

coefficient: material and temperature-dependent → K

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